| Item | Specitication |
| :---: | :---: |
| Measuring items | 1) Gas volume/internal gas pressure of carbonated beverages <br> 2) Air content and its oxygen concentration of carbonated beverages <br> 3) Inside gas pressure of non-carbonated beverages |
| Measuring range | $\begin{array}{ll}\text { 1) Pressure: } & 0.010 \text { to } 0.500 \mathrm{MPaG} \\ \text { 2) Temperature: } & 0 \text { to } 50^{\circ} \mathrm{C}\end{array}$ <br> 3) Residual gas volume: 0 to 30 mL <br> (Amount of residual gas after absorption of alkaline aqueous solution) <br> 4) Oxygen concentration: 0 to 21 vol\% (optional) (If the residual gas volume is less than 8 mL , measurement is not possible) |
| Accuracy | 1) Pressure: $\quad \pm 0.001 \mathrm{MPa}$ <br> 2) Temperature: $\quad \pm 0.05^{\circ} \mathrm{C}$ <br> 3) Residual gas volume: <br> $\pm 2 \%$ of the measured value (in case that the volume is 5 mL or more but less than 30 mL ) $\pm 0.1 \mathrm{~mL}$ of the measured value (in case that the volume is 0 mL or more but less than 5 mL ) (When measuring the standard gas specified by our company) <br> 4) Oxygen concentration: $\pm 0.2$ vol\% (optional) |
| Calibration | 1) Gas volume section Pressure sensor Calibration with a calibration pressure gauge <br> Zero calibration 0.000 MPaG <br> Span calibration 0.500 MPaG <br> 2) Gas volume section Temperature sensor Calibration with standard thermometer Offset adjustment approximately $20^{\circ} \mathrm{C}$ <br> 3) Air content section Pressure sensor Calibration with a calibration pressure gauge Zero calibration 0.000 MPaG Span calibration 0.050 MPaG <br> 4) Oxygen sensor Calibration by atmosphere and nitrogen gas Zero calibration 0.00 \% Span calibration 20.9 \% |
| Measurement time | Gas volume/internal gas pressure: Approx. 90 seconds for one sample <br> Residual gas volume: 5 to 12 minutes for one sample <br> *It depends on samples and conditions |
| Sample container | PET bottle:~2.0 ( 160 mL cans, various PET containers) <br> 1) For lightweight containers <br> 2) Compatible with tamper-resistant cap (double-layered lid) <br> 3) Maximum volume of container: <br> $\phi 110 \times 315 \mathrm{~mm}$ (height) |

## /EMKYOTO ELECTRONICS

 MANUFACTURING CO.,LTDhttps://www.kem.kyoto
Overseas Division : 2-7-1, Ichigaya-sadohara-cho, Shinjuku-ku Fax : +81-3-3268-5591 Phone: : +81-3-5227-3156

| Item | Specitication |
| :---: | :---: |
| Display | 16-digit x 1-line LCD display (with backlight) |
| Stored data | 500 latest measurement results 20 methods |
| Results obtained by calculation | 1) Gas volume from pressure and temperature <br> 2) Residual gas volume after absorption into alkaline aqueous solution <br> 3) Air content from residual gas volume and oxygen concentration <br> 4) Inside gas pressure of non-carbonated beverages |
| External output | 1) $\mathrm{RS}-232 \mathrm{C} 1 \mathrm{ch}$ <br> for connection to a printer or external computer <br> 2) USB port 1ch for data transfer or barcode reader |
| Ambient condition | 1) Temperature: 5 to $35^{\circ} \mathrm{C}$ ( 41 to $95^{\circ} \mathrm{F}$ ) <br> 2) Humidity: <br> $85 \% \mathrm{RH}$ or below (no condensation) |
| Power supply | AC 100 to $240 \mathrm{~V} \pm 10 \% 50 \mathrm{~Hz} / 60 \mathrm{~Hz}$ |
| Power consumption | 40 W |
| Coating | Alkali-resistant coating |
| External dimensions | 495(W) $\times 555(\mathrm{D}) \times 570(\mathrm{H}) \mathrm{mm}$ (19.5 (W) 21.8 (D) $\times 22.4$ (H) inches) |
| Weight | Approx. 38 kg (84 lbs) |
| Required air pressure | 0.6 to 0.7 MPaG |
| Option | 1) Dot Matrix Printer IDP-100 <br> 2) Oxygen Concentration Measurement Unit <br> 3) Data Acquisition Software SOFT-CAP |


3) SOFT-CAP

Gas Volume / Air content Analyzer

## GVA-71口



KYOTO ELECTRONICS
MANUFACTURING CO.,LTD.

## Summary / Features

## Measurement principle

## Summary

By simply setting a sample container on the sample stage, the following operations are performed automatically; opening, snifting, continuous shaking, the injection of the sample gas into the absorbent cylinder filled with alkaline solution, and the dissolution of the carbon dioxide gas in the absorbent cylinder.
By continuously rotating a sample container, the gas volume / internal gas pressure is calculated from the equilibrium pressure of the sample container and the sample temperature.
Furthermore, after absorbing the carbon dioxide gas in the absorbent cylinder, the amount of gas (residual gas) except for carbon dioxide is measured by the plunger pump and air content is calculated by measurement of oxygen concentration by the optional oxygen sensor.
Data printing is available by connecting an external printer.
 Output to a PC is also available.

## Features

Time reduction
Reduces pre-treatment and post-treatment time by
$30 \%$ and air volume measurement time by $70 \%$ compared to conventional instrument.

Various rinse mode
Multiple rinsing modes are available for each measurement mode
Convenient functions such as post-treatment and cleaning at the same time are provided.

## Method

Total of 20 methods can be set in an instrument.


Accuracy
A plunger pump is adopted to improve the air content measurement method.
Even low-volume air content can be measured more accurately than ever before.

Data management
Up to 500 measurement data stored in the main unit can be saved in the USB memory. The method can be created on the personal computer and imported into the instrument.

Barcode reader
By connecting a commercially available barcode reader, sample information can be easily transferred to the instrument, and sample information and measurement results can be associated and managed.


Gas volume
Calculate the gas volume from the conversion formula using the measured values of temperature and equilibrium pressure.
Selectable two types of conversion tables are programmed in the instrument, one for soft drinks and the other for beer, which can be switched by operators.

Air content
The amount of residual gas is calculated by a formula from the discharge amount and pressure of the plunger pump.
In addition, the oxygen concentration of the gas is measured, and it is converted into air content.

$$
\text { AIR = V × C / } 20.9
$$

AIR: air content $V$ : residual gas volume $C$ : oxygen concentration

Internal gas pressure
From the obtained gas volume and conversion formula, calculate the balance pressure when the sample temperature is $20^{\circ} \mathrm{C}$.

Inside gas pressure
Internal gas pressure of the sample at $20^{\circ} \mathrm{C}$ is calculated by the measured gas volume and conversion formula.

$$
\text { PRESS }=\frac{20+273.15}{\text { Temp }+273.15} \times \text { Press }
$$ Press: measured pressure Temp: sample temp

PRESS: equivalent pressure at $20^{\circ} \mathrm{C}$ (converted)

Measurement flow ※Measurement of gas volume and air content


